Periodic Table of the Elements

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Period	GROUP 1 IA	2 IIA	3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 VIII	9 VIII	10 VIII	11 IB	12 IIB	13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA
1	1 H Hydrogen 1.00794 $^2S_{1/2}$ 2.2 0.0899 13.5984 -259.14 -252.87 (v) 37 - $1s^1$ +1,-1																	2 He Helium 4.002602 $^1S_0$ - 0.1785 24.5874 - -268.93 (v) 32 - $1s^2$ 0
2	3 Li Lithium 6.941 $^2S_{1/2}$ 0.98 0.535 5.3917 180.54 1342 (m) 152 BCC [He] $2s^1$ +1	4 Be Beryllium 9.012182 $^1S_0$ 1.57 1.848 9.3227 1287 2470 (m) 112 HCP [He] $2s^2$ +2											5 B Boron 10.811 $^2P°_{1/2}$ 2.04 2.46 8.2980 2075 4000 (v) 82 rhom. [He] $2s^2\,2p^1$ +3	6 C Carbon 12.0107 $^3P_0$ 2.55 2.26 11.2603 3550 4027 (v) 77 hex [He] $2s^2\,2p^2$ +2,4,-4	7 N Nitrogen 14.0067 $^4S°_{3/2}$ 3.04 1.251 14.5341 -210.1 -195.79 (v) 75 - [He] $2s^2\,2p^3$ +2,3,4,5,-2,-3	8 O Oxygen 15.9994 $^3P_2$ 3.44 1.429 13.6181 -218.3 -182.9 (v) 73 - [He] $2s^2\,2p^4$ -2	9 F Fluorine 18.9984032 $^2P°_{3/2}$ 3.98 1.696 17.4228 -219.6 -188.12 (v) 71 - [He] $2s^2\,2p^5$ -1	10 Ne Neon 20.1797 $^1S_0$ - 0.9 21.5645 -248.59 -246.08 (v) 69 - [He] $2s^2\,2p^6$ 0
3	11 Na Sodium 22.989770 $^2S_{1/2}$ 0.93 0.968 5.1391 97.72 883 (m) 186 BCC [Ne] $3s^1$ +1	12 Mg Magnesium 24.3050 $^1S_0$ 1.31 1.738 7.6462 650 1090 (m) 160 HCP [Ne] $3s^2$ +2											13 Al Aluminum 26.981538 $^2P°_{1/2}$ 1.61 2.7 5.9858 660.32 2519 (m) 143 FCC [Ne] $3s^2\,3p^1$ +3	14 Si Silicon 28.0855 $^3P_0$ 1.90 2.33 8.1517 1414 2900 (v) 111 cubic [Ne] $3s^2\,3p^2$ +2,4,-4	15 P Phosphorus 30.97361 $^4S°_{3/2}$ 2.19 1.823 10.4867 44.2 280.5 (v) 106 § [Ne] $3s^2\,3p^3$ +3,4,5,-3	16 S Sulfur 32.065 $^3P_2$ 2.58 1.96 10.3600 115.21 444.72 (v) 102 FCO [Ne] $3s^2\,3p^4$ +2,4,6,-2	17 Cl Chlorine 35.453 $^2P°_{3/2}$ 3.16 3.214 12.9676 -101.5 -34.04 (v) 99 - [Ne] $3s^2\,3p^5$ +1,3,5,7,-1	18 Ar Argon 39.948 $^1S_0$ - 1.784 15.7596 -189.3 -185.8 (v) 97 - [Ne] $3s^2\,3p^6$ 0
4	19 K Potassium 39.0983 $^2S_{1/2}$ 0.82 0.856 4.3407 63.38 759 (m) 227 BCC [Ar] $4s^1$ +1	20 Ca Calcium 40.078 $^1S_0$ 1.00 1.55 6.1132 842 1484 (m) 197 FCC [Ar] $4s^2$ +2	21 Sc Scandium 44.955910 $^2D_{3/2}$ 1.36 2.985 6.5615 1541 2830 (m) 162 HCP [Ar] $3d^1\,4s^2$ +3	22 Ti Titanium 47.867 $^3F_2$ 1.54 4.507 6.8281 1668 3287 (m) 147 HCP [Ar] $3d^2\,4s^2$ +2,3,4	23 V Vanadium 50.9415 $^4F_{3/2}$ 1.63 6.11 6.7462 1910 3407 (m) 134 BCC [Ar] $3d^3\,4s^2$ +2,3,4,5	24 Cr Chromium 51.9961 $^7S_3$ 1.66 7.14 6.7665 1907 2671 (m) 128 BCC [Ar] $3d^5\,4s^1$ +2,3,6	25 Mn Manganese 54.938049 $^6S_{5/2}$ 1.55 7.47 7.4340 1246 2061 (m) 127 §cubic [Ar] $3d^5\,4s^2$ +2,3,4,6,7	26 Fe Iron 55.845 $^5D_4$ 1.83 7.874 7.9024 1538 2861 (m) 126 BCC [Ar] $3d^6\,4s^2$ +2,3	27 Co Cobalt 58.933200 $^4F_{9/2}$ 1.88 8.9 7.8810 1495 2927 (m) 125 HCP [Ar] $3d^7\,4s^2$ +2,3	28 Ni Nickel 58.6934 $^3F_4$ 1.91 8.908 7.6398 1455 2913 (m) 124 FCC [Ar] $3d^8\,4s^2$ +2,3	29 Cu Copper 63.546 $^2S_{1/2}$ 1.90 8.92 7.7264 1084.62 2927 (m) 128 FCC [Ar] $3d^{10}\,4s^1$ +1,2	30 Zn Zinc 65.409 $^1S_0$ 1.65 7.14 9.3942 419.53 907 (m) 134 §hex [Ar] $3d^{10}\,4s^2$ +2	31 Ga Gallium 69.723 $^2P°_{1/2}$ 1.81 5.904 5.9993 29.76 2204 (m) 135 §BCO [Ar] $3d^{10}\,4s^2\,4p^1$ +3	32 Ge Germanium 72.64 $^3P_0$ 2.01 5.323 7.8994 938.3 2820 (v) 122 §cubic [Ar] $3d^{10}\,4s^2\,4p^2$ +2,4	33 As Arsenic 74.92160 $^4S°_{3/2}$ 2.18 5.727 9.7886 817 614 (v) 119 rhom. [Ar] $3d^{10}\,4s^2\,4p^3$ +3,5,-3	34 Se Selenium 78.96 $^3P_2$ 2.55 4.819 9.7524 221 685 (v) 116 §hex [Ar] $3d^{10}\,4s^2\,4p^4$ +2,4,6,-2	35 Br Bromine 79.904 $^2P°_{3/2}$ 2.96 3.12 11.8138 -7.3 59 (v) 114 BCO [Ar] $3d^{10}\,4s^2\,4p^5$ +1,5,-1	36 Kr Krypton 83.798 $^1S_0$ 3 3.75 13.9996 -157.36 -153.22 (v) 110 - [Ar] $3d^{10}\,4s^2\,4p^6$ 0
5	37 Rb Rubidium 85.4678 $^2S_{1/2}$ 0.82 0.532 4.1771 39.31 688 (m) 248 BCC [Kr] $5s^1$ +1	38 Sr Strontium 87.62 $^1S_0$ 0.95 2.63 5.6949 777 1382 (m) 215 FCC [Kr] $5s^2$ +2	39 Y Yttrium 88.90585 $^2D_{3/2}$ 1.22 4.472 6.2173 1526 3345 (m) 180 HCP [Kr] $4d^1\,5s^2$ +3	40 Zr Zirconium 91.224 $^3F_2$ 1.33 6.511 6.6339 1855 4409 (m) 160 HCP [Kr] $4d^2\,5s^2$ +4	41 Nb Niobium 92.90638 $^6D_{1/2}$ 1.60 8.57 6.7589 2477 4744 (m) 146 BCC [Kr] $4d^4\,5s^1$ +3,5	42 Mo Molybdenum 95.94 $^7S_3$ 2.16 10.28 7.0924 2623 4639 (m) 139 BCC [Kr] $4d^5\,5s^1$ +2,3,4,5,6	43 Tc Technetium (98) $^6S_{5/2}$ 1.9 11.5 7.28 2157 4265 (m) 136 HCP [Kr] $4d^5\,5s^2$ +4,7	44 Ru Ruthenium 101.07 $^5F_5$ 2.20 12.37 7.3605 2334 4150 (m) 134 HCP [Kr] $4d^7\,5s^1$ +2,3,4,6,8	45 Rh Rhodium 102.90550 $^4F_{9/2}$ 2.28 12.45 7.4589 1964 3695 (m) 134 FCC [Kr] $4d^8\,5s^1$ +2,3,4	46 Pd Palladium 106.42 $^1S_0$ 2.20 12.023 8.3369 1554.9 2963 (m) 137 FCC [Kr] $4d^{10}$ +2,4	47 Ag Silver 107.8682 $^2S_{1/2}$ 1.93 10.49 7.5762 961.78 2162 (m) 144 FCC [Kr] $4d^{10}\,5s^1$ +1	48 Cd Cadmium 112.411 $^1S_0$ 1.69 8.65 8.9938 321.07 767 (m) 151 §hex [Kr] $4d^{10}\,5s^2$ +2	49 In Indium 114.818 $^2P°_{1/2}$ 1.78 7.31 5.7864 156.6 2072 (m) 167 §tetra. [Kr] $4d^{10}\,5s^2\,5p^1$ +3	50 Sn Tin 118.710 $^3P_0$ 1.96 7.31 7.3439 231.93 2602 (v) 141 §tetra. [Kr] $4d^{10}\,5s^2\,5p^2$ +2,4	51 Sb Antimony 121.760 $^4S°_{3/2}$ 2.05 6.697 8.6084 630.63 1587 (v) 138 §rhom. [Kr] $4d^{10}\,5s^2\,5p^3$ +3,5,-3	52 Te Tellurium 127.60 $^3P_2$ 2.10 6.24 9.0096 449.51 988 (v) 135 hex [Kr] $4d^{10}\,5s^2\,5p^4$ +2,4,6,-2	53 I Iodine 126.90447 $^2P°_{3/2}$ 2.66 4.94 10.4513 113.7 184.3 (v) 133 BCO [Kr] $4d^{10}\,5s^2\,5p^5$ +1,5,7,-1	54 Xe Xenon 131.293 $^1S_0$ 2.60 5.9 12.1298 -111.8 -108 (v) 130 - [Kr] $4d^{10}\,5s^2\,5p^6$ 0
6	55 Cs Cesium 132.90545 $^2S_{1/2}$ 0.79 1.879 3.8939 28.44 671 (m) 265 BCC [Xe] $6s^1$ +1	56 Ba Barium 137.327 $^1S_0$ 0.89 3.51 5.2117 727 1870 (m) 222 BCC [Xe] $6s^2$ +2	Lanthanide Series	72 Hf Hafnium 178.49 $^3F_2$ 1.3 13.31 6.8251 2233 4603 (m) 159 HCP [Xe] $4f^{14}\,5d^2\,6s^2$ +4	73 Ta Tantalum 180.9479 $^4F_{3/2}$ 1.5 16.65 7.5496 3017 5458 (m) 146 BCC [Xe] $4f^{14}\,5d^3\,6s^2$ +5	74 W Tungsten 183.84 $^5D_0$ 2.36 19.25 7.8640 3422 5555 (m) 139 BCC [Xe] $4f^{14}\,5d^4\,6s^2$ +2,3,4,5,6	75 Re Rhenium 186.207 $^6S_{5/2}$ 1.9 21.02 7.8335 3186 5596 (m) 137 HCP [Xe] $4f^{14}\,5d^5\,6s^2$ +2,4,6,7,-1	76 Os Osmium 190.23 $^5D_4$ 2.2 22.61 8.4382 3033 5012 (m) 135 HCP [Xe] $4f^{14}\,5d^6\,6s^2$ +2,3,4,6,8	77 Ir Iridium 192.217 $^4F_{9/2}$ 2.2 22.65 8.9670 2466 4428 (m) 136 FCC [Xe] $4f^{14}\,5d^7\,6s^2$ +2,3,4,6	78 Pt Platinum 195.078 $^3D_3$ 2.28 21.09 8.9588 1768.3 3825 (m) 139 FCC [Xe] $4f^{14}\,5d^8\,6s^1$ +2,4	79 Au Gold 196.96655 $^2S_{1/2}$ 2.54 19.3 9.2255 1064.18 2856 (m) 144 FCC [Xe] $4f^{14}\,5d^{10}\,6s^1$ +1,3	80 Hg Mercury 200.59 $^1S_0$ 2 13.534 10.4375 -38.83 356.73 (m) 151 §rhom. [Xe] $4f^{14}\,5d^{10}\,6s^2$ +1,2	81 Tl Thallium 204.3833 $^2P°_{1/2}$ 1.62 11.85 6.1082 304 1473 (m) 170 HCP [Hg] $6p^1$ +1,3	82 Pb Lead 207.2 $^3P_0$ 2.33 11.34 7.4167 327.46 1749 (m) 175 FCC [Hg] $6p^2$ +2,4	83 Bi Bismuth 208.98038 $^4S°_{3/2}$ 2.02 9.78 7.2855 271.3 1564 (v) 146 §rhom. [Hg] $6p^3$ +3,5	84 Po Polonium (209) $^3P_2$ 2.0 9.196 8.414 254 962 - §cubic [Hg] $6p^4$ +2,4	85 At Astatine (210) $^2P°_{3/2}$ 2.2 - - 302 - - - [Hg] $6p^5$ +1,3,5,7,-1	86 Rn Radon (222) $^1S_0$ - 9.73 10.7485 -71 -61.7 (v) 145 - [Hg] $6p^6$ 0
7	87 Fr Francium (223) $^2S_{1/2}$ 0.7 - 4.0727 - - - - [Rn] $7s^1$ +1	88 Ra Radium (226) $^1S_0$ 0.9 5 5.2784 700 1737 - BCC [Rn] $7s^2$ +2	Actinide Series	104 Rf Rutherfordium (261) $^3F_2$? 6.0 ? [Rn] $5f^{14}\,6d^2\,7s^2$ +4	105 Db Dubnium (262) [Rn] $5f^{14}\,6d^3\,7s^2$	106 Sg Seaborgium (266) [Rn] $5f^{14}\,6d^4\,7s^2$	107 Bh Bohrium (264) [Rn] $5f^{14}\,6d^5\,7s^2$	108 Hs Hassium (277) [Rn] $5f^{14}\,6d^6\,7s^2$	109 Mt Meitnerium (268) [Rn] $5f^{14}\,6d^7\,7s^2$?	110 Ds Darmstadtium (281) [Rn] $5f^{14}\,6d^8\,7s^2$?	111 Rg Roentgenium (272) [Rn] $5f^{14}\,6d^9\,7s^2$?	112 Cn Copernicium (285) [Rn] $5f^{14}\,6d^{10}\,7s^2$?	113 Nh Nihonium [Rn]$5f^{14}6d^{10}7s^27p^1$?	114 Uuq Ununquadium (289) [Rn]$5f^{14}6d^{10}7s^27p^2$?	115 Mc Moscovium [Rn]$5f^{14}6d^{10}7s^27p^3$?	116 Lv Livermorium (292) [Rn]$5f^{14}6d^{10}7s^27p^4$?	117 Ts Tennessine [Rn]$5f^{14}6d^{10}7s^27p^5$?	118 Og Oganesson [Rn]$5f^{14}6d^{10}7s^27p^6$?

Phase at STP

Gas Liquid Solid Synthetic

Categories

Alkali Metals	Noble Gas
Alkaline Earth Metals	Halogens
Transition Metals	Non Metals
Rare Earth Metals	
Poor Metals	Metalloids

Atomic Number
Symbol
Name
*Density [Note]
*Melting Point (°C)
Atomic radius (pm)[Note]

†Atomic Weight
†Ground-State Level
*Electronegativity (Pauling)
†Ionization Energy (eV)
*Boiling Point (°C)
Crystal Structure [Note]
†Electron Configuration
Possible Oxidation States [Note]

†Common Constants

Source: physics.nist.gov

Absolute Zero		-273.15 °C	Gravitation Constant	G	6.67428×10^{-11} $m^3\,kg^{-1}\,s^{-2}$
Atomic Mass Unit	m_u	1.660539×10^{-27} kg	Molar Gas Constant	R	8.314472 J $mol^{-1}\,K^{-1}$
Avogadro Constant		6.022142×10^{23} mol^{-1}	Molar Volume (Ideal Gas)		0.02241410 m^3/mol
Base of Natural Logarithms	e	2.718281828	PI	π	3.14159265358979
Boltzmann constant	k	1.380650×10^{-23} J/K	Planck Constant	h	6.626069×10^{-34} J s
Electron Mass	m_e	9.10938215×10^{-31} kg	Proton-Electron Mass Ratio	m_p/m_e	1836.15267**389**
	$m_e c^2$	0.5110 MeV	Rydberg Constant	R_∞	10 973 732 m^{-1}
Electron Radius (Classical)	r_e	2.8179403×10^{-15} m		$R_\infty c$	3.289842×10^{15} Hz
Electron Volt	eV	1.602176×10^{-19} J		$R_\infty hc$	13.6057 eV
Elementry Charge	e	1.602176×10^{-19} C	Second Radiation Constant	ch/k	0.01438769 m K
Faraday Constant	F	96 485.3399 C/mol	Speed of Light in a Vacuum	c	299 792 458 m/s
fine-structure constant	α	0.0072973525	Speed of sound in air at STP		343.2 m/s
First Radiation Constant	$2\pi hc^2$	3.7417749×10^{-16} W m^2	Standard Pressure		101 325 Pa

Lanthanides	57 La Lanthanum 138.9055 $^2D_{3/2}$ 1.10 6.146 5.5769 920 3464 (m) 187 §hex [Xe] $5d^1\,6s^2$ +3	58 Ce Cerium 140.116 $^1G°_4$ 1.12 6.689 5.5387 798 3360 (m) 182 FCC [Xe] $4f^1\,5d^1\,6s^2$ +3,4	59 Pr Praseodymium 140.90765 $^4I°_{9/2}$ 1.13 6.64 5.473 931 3290 (m) 182 §hex [Xe] $4f^3\,6s^2$ +3,4	60 Nd Neodymium 144.24 $^5I_4$ 1.14 7.01 5.5250 1021 3100 (m) 181 §hex [Xe] $4f^4\,6s^2$ +3	61 Pm Promethium (145) $^6H°_{5/2}$ - 7.264 5.582 1100 3000 (m) 183 HCP [Xe] $4f^5\,6s^2$ +3	62 Sm Samarium 150.36 $^7F_0$ 1.17 7.353 5.6437 1072 1803 (m) 180 §hex [Xe] $4f^6\,6s^2$ +2,3	63 Eu Europium 151.964 $^8S°_{7/2}$ - 5.244 5.6704 822 1527 (m) 180 BCC [Xe] $4f^7\,6s^2$ +2,3	64 Gd Gadolinium 157.25 $^9D°_2$ 1.20 7.901 6.1498 1313 3250 (m) 180 HCP [Xe] $4f^7\,5d^1\,6s^2$ +3	65 Tb Terbium 158.92534 $^6H°_{15/2}$ - 8.219 5.8638 1356 3230 (m) 177 HCP [Xe] $4f^9\,6s^2$ +3,4	66 Dy Dysprosium 162.500 $^5I_8$ 1.22 8.551 5.9389 1412 2567 (m) 178 HCP [Xe] $4f^{10}\,6s^2$ +3	67 Ho Holmium 164.93032 $^4I°_{15/2}$ 1.23 8.795 6.0215 1474 2700 (m) 176 HCP [Xe] $4f^{11}\,6s^2$ +3	68 Er Erbium 167.259 $^3H_6$ 1.24 9.066 6.1077 1497 2868 (m) 176 HCP [Xe] $4f^{12}\,6s^2$ +3	69 Tm Thulium 168.93421 $^2F°_{7/2}$ 1.25 9.321 6.1843 1545 1950 (m) 176 HCP [Xe] $4f^{13}\,6s^2$ +2,3	70 Yb Ytterbium 173.04 $^1S_0$ - 6.57 6.2542 819 1196 (m) 176 FCC [Xe] $4f^{14}\,6s^2$ +2,3	71 Lu Lutetium 174.967 $^2D_{3/2}$ 1.27 9.841 5.4259 1663 3402 (m) 174 HCP [Xe] $4f^{14}\,5d^1\,6s^2$ +3
Actinides	89 Ac Actinium (227) $^2D_{3/2}$ 1.1 10.07 5.17 1050 3200 - FCC [Rn] $6d^1\,7s^2$ +3	90 Th Thorium 232.0381 $^3F_2$ 1.3 11.724 6.3067 1750 4820 (m) 179 FCC [Rn] $6d^2\,7s^2$ +4	91 Pa Protactinium 231.0359 $^4K_{11/2}$ 1.5 15.37 5.89 1572 4000 (m) 163 §tetra [Rn] $5f^2\,6d^1\,7s^2$ +4,5	92 U Uranium 238.0289 $^5L°_6$ 1.38 19.05 6.1941 1135 3927 (m) 156 BCP [Rn] $5f^3\,6d^1\,7s^2$ +3,4,5,6	93 Np Neptunium (237) $^6L°_{11/2}$ 1.36 20.45 6.2657 644 4000 (m) 155 SO [Rn] $5f^4\,6d^1\,7s^2$ +3,4,5,6	94 Pu Plutonium (244) $^7F_0$ 1.28 19.816 6.0260 640 3230 (m) 159 §mono. [Rn] $5f^6\,7s^2$ +3,4,5,6	95 Am Americium (243) $^8S°_{7/2}$ 1.3 - 5.9738 1176 2011 (m) 173 HCP [Rn] $5f^7\,7s^2$ +3,4,5,6	96 Cm Curium (247) $^9D°_2$ 1.3 13.51 5.9914 1345 3110 (m) 174 HCP [Rn] $5f^7\,6d^1\,7s^2$ +3	97 Bk Berkelium (247) $^6H°_{15/2}$ 1.3 14.78 6.1979 1050 - (m) 170 hex [Rn] $5f^9\,7s^2$ +3,4	98 Cf Californium (251) $^5I_8$ 1.3 15.1 6.2817 900 - - hex [Rn] $5f^{10}\,7s^2$ +3	99 Es Einsteinium (252) $^4I°_{15/2}$ 1.3 - 6.42 860 - - - [Rn] $5f^{11}\,7s^2$ +3	100 Fm Fermium (257) $^3H_6$ 1.3 - 6.50 1527 - - - [Rn] $5f^{12}\,7s^2$ +3	101 Md Mendelevium (258) $^2F°_{7/2}$ 1.3 - 6.58 827 - - - [Rn] $5f^{13}\,7s^2$ +2,3	102 No Nobelium (259) $^1S_0$ 1.3 - 6.65 827 - - - [Rn] $5f^{14}\,7s^2$ +2,3	103 Lr Lawrencium (262) $^2P°_{1/2}$? - 4.9 ? 1627 - - - [Rn] $5f^{14}\,7s^2\,7p^1$ +3

Notes:
- Density units are g/cm^3 for solids and g/L or kg/m^3 at 0° Celsius for gases
- Atomic Weight based on ^{12}C
- ( ) indicate mass number of most stable isotope
- Common Oxidation States in bold
- Electron Config. based on IUPAC guidelines
- § indicates crystal structure is unusual or may require explanation
- (m) Metallic radius, (v) Covalent radius

References:
†Nist.gov, *Wolfram.com (Mathematic), CRC Handbook of Chemistry and Physics 81st Edition, 2000-2001, and others

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